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Effect of rhodium doping on the photorefractive properties of BCT crystals at 850 nm

S.Bernhardt⁽¹⁾, H.Veenhuis⁽²⁾, P.Delays⁽¹⁾, R.Pankrath⁽²⁾, G.Roosen⁽¹⁾

⁽¹⁾ Laboratoire Charles Fabry de l'Institut d'Optique, Unité Mixte de Recherche 8501 du Centre National de la Recherche Scientifique, Bat. 503, Centre Scientifique d'Orsay, B.P. 147, 91403 Orsay Cedex, France.

⁽²⁾ Universität Osnabrück, Fachbereich Physik, D 49069 Osnabrück, Germany

Corresponding Author : Ph. Delays, Tel : 33 1 69358750, Fax : 33 1 69358700, Email :
Philippe.delays@iota.u-psud.fr

Abstract : We present an experimental investigation of the photorefractive properties of rhodium doped barium calcium titanate crystals of the congruently melting composition $\text{Ba}_{0.77}\text{Ca}_{0.23}\text{TiO}_3$. Considering the results previously obtained on this crystal in the visible region, it should be a good alternative to BaTiO_3 . Nevertheless, many applications use infrared light. Therefore we present here a study of rhodium doped BCT crystals at 850 nm. This wavelength is of special interest as it is in the spectral range of laser diodes. Rhodium doping is expected to enhance the sensitivity of the crystal in the infrared as it is the case for BaTiO_3 .

We first noticed that BCT:Rh crystals are sensitive at this wavelength as expected. Furthermore, the photorefractive properties are interesting in term of photorefractive gain, with a gain as high as 3 cm^{-1} with ordinary polarization. This study has also put forward the fact that rhodium is not the only defect that participates in the photorefractive effect. Indeed, a large quantity of iron seems to be present in the BCT crystals and to participate in the photorefractive effect.

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It has been demonstrated that rhodium improves the infrared sensitivity of BaTiO₃ so that the crystal is of special interest for many applications. It exhibits high photorefractive gain, small response time and its behaviour is well explained with a two-level model with three states of charges of rhodium : Rh³⁺, Rh⁴⁺ and Rh⁵⁺ [1, 2]. But it suffers from a drawback that prevents its use in industrial systems : a phase transition around 5°C, that can destroy it .

Recently, first results about a new ferroelectric crystal derived from BaTiO₃ with no phase transition at room temperature, Ba_{0.77}Ca_{0.23}TiO₃, were published and show that this crystal has high electrooptic coefficients and high gain in the visible region [3--5]. As BCT is close to BaTiO₃, it is expected that rhodium will also improve the sensitivity of BCT in the infrared. BCT:Rh may thus be an alternative to BaTiO₃:Rh, provided its photorefractive properties in terms of response time, photorefractive gain and saturation intensity are good enough. Therefore, we will focus in this paper on the photorefractive properties of this new material at the near infrared wavelength 850 nm. Absorption and two-beam coupling gain measurements are carried out in order to find out whether rhodium doping improves the photorefractive performances of BCT in the infrared.

1. EXPERIMENTAL RESULTS

1.1. Crystals

BCT crystals grown with 0, 220, 370, 600, 1000 and 2000 ppm of rhodium added in the melt and one crystal with 140 ppm of iron are studied. Typical dimensions are 2*3*4 (=c) mm³.

The absorption spectra are presented in Fig.1 (except for the 600 ppm rhodium doped crystal for which the absorption spectrum was not measured). The influence of rhodium on absorption is the same as in BaTiO₃:Rh with a large absorption band around 600 nm which was attributed to this dopant [5, 6]. We can note in Fig.1 that the absorption band seems to increase proportionally to the amount of rhodium added in the melt. To confirm this point, we normalized the absorption spectra to the quantity of rhodium added in the melt (Fig.2a). To improve this comparison, we first substrated the absorption contribution of the undoped crystal from the ones of the rhodium doped samples as we consider the absorption of the undoped BCT as a residual absorption present in all crystals. A departure from this assumption can explain the discrepancy between the curves in the short wavelength part of the spectrum (400-600 nm). In the upper part of the spectrum, the low amplitude of the absorption background leads to a minor influence of the assumption except for crystals containing small amount of rhodium (like the 200 ppm rhodium doped crystal for which the calculation is not performed). The similarity of the normalized absorption curves leads us to the conclusion that the observed absorption band is proportional to the rhodium doping.

Moreover, we also see that the absorption spectra for the undoped and iron doped crystals are similar. Indeed, as shown by Fig.2b, both spectra are proportional, which certainly indicates the presence of iron in the undoped crystal.

1.2. Two beam coupling experimental procedure

We carried out measurements of the photorefractive gain by using the two beam coupling experimental arrangement shown in Fig.3. Two coherent beams issued from a DBR laser diode at 850 nm were aligned so that the direction of the grating wave vector made an angle β with the c-axis of the crystal. All the experiments (with the exception of the measurement of r_{42}^T) are performed with ordinary polarizations and with symmetric incidences of the beams ($\beta=0$). The choice of the ordinary polarization enables us to measure moderate energy transfers and thus prevents us from any measurement problems due to depletion of the pump beam, beam fanning or oscillation of beams inside the crystal that often occurs with extraordinary polarizations. The signal beam diameter was reduced compared to the one of the pump beam by passing through a hole H to ensure a good overlap between the two beams. Furthermore, the intensity of the signal beam was much less than the one of the pump beam (a factor of 100) in order to be in the undepleted pump approximation. From the steady state intensities of the transmitted signal beam without (I_s) and with (I_{sp}) illumination by the pump beam, we calculated the two beam coupling coefficient Γ according to Eq. 1 :

$$\Gamma = \frac{1}{l} \ln \left(\frac{I_{sp}}{I_s} \right) \quad (1)$$

where l is the length of interaction between the two beams, which corresponds here to the length of the crystal. Each measurement datum corresponds to a measurement in attenuation and in amplification allowing to separate the photorefractive components from an eventual absorption contribution (induced absorption or absorption grating). We measured the dependence of the gain with several parameters : intensity of the pump beam, grating wave vector, angle between the c-axis and the grating wave vector β .

1.3. Theoretical aspect

The dependence of the gain with the grating wave vector k_r and with intensity I can be most of the time described by a general expression of the form [7, 8, 9, 10] :

$$\Gamma = \frac{2\pi}{\lambda_0} n^3 \frac{k_B T}{e} r^{\text{eff}} \eta(I) \frac{k_r}{1 + \frac{k_r^2}{k_0^2(I)}} \quad (2)$$

where λ_0 is the wavelength of light in vacuum, n the mean refractive index and r^{eff} the effective electrooptic coefficient.

In this expression, $\eta(I)$ expresses the saturation of the gain with intensity which generally writes as :

$$\eta(I) = \frac{1}{1 + \frac{I_{\text{sat}}}{I}} \quad (3)$$

and k_0 expresses as a function of the effective trap density N_{eff} according to (with the effective dielectric constant $\epsilon^{\text{eff}}(\beta)$):

$$k_0^2(I) = \frac{e^2 N_{\text{eff}}(I)}{k_B T \epsilon_0 \epsilon^{\text{eff}}(\beta)} \quad (4)$$

On the one hand, the saturation of the gain is physically linked to the respective amplitudes of the photoconductivity and the dark conductivity, $\eta(I)$ going to one for illumination at which photoconductivity overpasses the dark conductivity. On the other hand, the intensity dependence of the effective trap density is due to the charge redistributions that can occur among different photorefractive species. Using these expressions (Eq. 1 to 4) to adjust our experimental results, we will deduce three characteristic parameters of the crystals : the effective electrooptic coefficient, the saturation intensity and the effective trap density. These parameters will enable to compare the different crystals together.

1.4. Experimental results

1.4.a. Gain as a function of intensity

The intensity dependence of the gain gives information about the intensity necessary to saturate the gain. It is a very important parameter for some applications like phase conjugation that have a photorefractive gain threshold and where a decrease of the gain can prevent from passing this threshold. The intensity dependence of the gain will allow to determine the power of the laser source that will be necessary to implement such a device. This problem has especially to be considered in the infrared because the photoconductivity of materials like BCT decreases.

We fixed the grating spacing at $1.5 \mu\text{m}^{-1}$, what implies that we have $k_r^2 \ll k_0^2$, leading to a dependence given by :

$$\Gamma(I) = \eta(I) \Gamma_{\infty} = \frac{\Gamma_{\infty}}{1 + \frac{I_{\text{sat}}}{I}} \quad (5)$$

where $\Gamma_{\infty} = \frac{2\pi}{\lambda_0} n^3 r^{\text{eff}} \frac{k_B T}{e} k_r$ is the saturated gain. Experimentally, I_{sat} is the intensity necessary to reach half of the maximum gain.

By fitting the experimental results according to Eq. 5 (Fig.4a), we deduce the values of the saturation intensity for each crystal (Table 1). In Fig.4a, the saturated gain Γ_{∞} is normalized to 1 in order to enable the comparison of the curves. First, we noticed that the saturation intensity decreases when the quantity of rhodium in the crystals goes from 0 to 2000 ppm. Furthermore, the saturation intensity of the iron doped crystal is the highest. Second, we noticed that the saturation intensity for the BCT:Rh crystals is between 100 and 1000 time higher than for BaTiO₃:Rh. Indeed, we measured the saturation intensity for a BaTiO₃:Rh crystal with 2000 ppm of rhodium (labelled Y32-B) and we found 0.5

mW.cm⁻². It means that we need 100 to 1000 times more intensity to saturate the gain of a BCT:Rh crystal than of a BaTiO₃:Rh crystal if the crystals are doped with the same amount of rhodium.

1.4.b. Gain as a function of the grating wave vector

Measurements of the gain as a function of the grating wave vector are carried out to determine the effective electrooptic coefficient and the effective trap density of the crystals. The measurements are performed with an intensity sufficient to saturate the gain, determined according to the previous measurements. Then, according to Eq. 2, the photorefractive gain expresses as :

$$\Gamma = \Gamma_{\infty} = \frac{2\pi}{\lambda_0} n^3 \frac{k_B T}{e} r^{\text{eff}} \frac{k_r}{1 + \frac{k_r^2}{k_0^2}} \quad (6)$$

Two kinds of measurements were performed. First, using the arrangement shown in Fig. 3, we measured the two beam coupling gain with copropagating waves. In this configuration, for $\beta=0$, we performed measurements at small grating wave vector ($k_r^2 \ll k_0^2$). According to Eq.6, we have :

$$\Gamma \approx \frac{2\pi}{\lambda_0} n^3 r^{\text{eff}} \frac{k_B T}{e} k_r \quad (7)$$

The linear dependence of the gain with the grating wave vector enables to determine the effective electrooptic coefficient but will not give a precise value of the effective trap density, which appears as a departure from the linear curve at high value of k_r .

To determine the effective trap density, we carried out another measurement in the configuration presented on Fig. 5, i.e. with counterpropagating beams [7]. In this particular case, with a large grating wave vector ($k_r^2 \gg k_0^2$), the gain is dominated by k_0^2 (i.e. the trap limited field E_q defined in a one trap model [10]) and this measurement enables to evaluate the effective trap density according to Eq. 4.

No experimental points are given between 7 and 30 μm^{-1} because these values of grating wave vector are either forbidden due to refraction laws (14-30 μm^{-1}) with the crystal cut we used, or correspond to large angle between the beams, what leads to problem for the beams to overlap in the whole crystal (7-14 μm^{-1}). Nevertheless, the conjugation of copropagative ($k < 10 \mu\text{m}^{-1}$) and counterpropagating ($k \approx 30 \mu\text{m}^{-1}$) measurements is sufficient to determine precisely the effective trap density [7].

We performed then a simultaneous fit with two parameters on the measurements with copropagating and counterpropagating waves (Fig.4b) using a function of type $\Gamma = A r^{\text{eff}} \frac{k_r}{1 + \frac{k_r^2}{k_0^2}}$ (A is a known

constant). We obtained the effective electrooptic coefficient r^{eff} and the parameter k_0^2 from which we deduced the effective trap density (Eq. 4) using the value of $\epsilon^{\text{eff}}(0)$ calculated in [4] ($\epsilon^{\text{eff}}(0) = 181$). In the configuration of the experiment, the effective electrooptic coefficient is $r_{22}^{\text{eff}}(0)$ and expresses as :

$$r_{22}^{\text{eff}}(0) = r_{13}^S + \frac{p_{13}^E e_{33}}{c_{33}^E} \quad (8)$$

if we consider the indirect electrooptic contributions (elasto-optic, elastic and piezoelectric) [4].

In a simpler way, the effective trap density is deduced from the maximum of the curves presented in Fig. 4b and the effective electrooptic coefficient is given by the slope of the curves at small grating wave vectors. From the fit, we obtained an almost constant value $N_{\text{eff}} = (6 \pm 2) \cdot 10^{22} \text{ m}^{-3}$ whatever the crystals (Table 1), whether they were rhodium doped from 200 to 2000 ppm, iron doped or undoped. The mean value of the electrooptic coefficient is $r_{22}^{\text{eff}} = (13.5 \pm 1.3) \text{ pm} \cdot \text{V}^{-1}$ (Table 1).

1.4.c. Dependence of the effective trap density on intensity

We also measured the dependence of the effective trap density on intensity on a 1000 ppm rhodium doped crystal. The experimental conditions are similar as for the previous measurements presented in 1.4.b. This time, we carried out measurements of the gain as a function of the grating wave vector for several intensities. These intensities have been chosen in a range so that the gain depends on intensity. We chose intensities of $2 \text{ W} \cdot \text{cm}^{-2}$, $200 \text{ mW} \cdot \text{cm}^{-2}$ and $50 \text{ mW} \cdot \text{cm}^{-2}$. We see in figure 6 that the position of the maximum of the curves, which corresponds to the effective trap density, depends on intensity in the chosen range. Adjustements performed as explained in 1.4.b enable us to determine the effective trap density for each laser intensity. The results presented in table 2 show a dependence of the effective trap density on intensity.

1.4.d. Determination of r_{42}^T

The experimental configuration for the determination of the unclamped value of the electrooptic coefficient r_{42}^T is presented in Fig.3 with a crystal oriented such as the grating wave vector is perpendicular to the c-axis. The beams are extraordinary polarized. We performed measurements of the gain as a function of β around $\beta = \frac{\pi}{2}$ with $k_r = 0.8 \mu\text{m}^{-1}$ and we deduced the value of r_{42}^T from the slope of the experimental curve according to a procedure presented in [4]. These experiments were carried out on two crystals : BCT:Rh 1000 ppm and BCT:Rh 2000 ppm. The results are presented in Fig. 7. We found a value of 276 pm/V for the 1000 ppm rhodium doped crystal and 215 pm/V for the 2000 ppm rhodium doped one. Taking into account the accuracy of the measurement, these values are in good accordance with the value of $250 \text{ pm} \cdot \text{V}^{-1}$ measured on BCT:Rh 100 ppm at 532 nm [4].

1.5. Discussion

The aim of these experiments was to determine whether rhodium improves the photorefractive properties of BCT crystals at 850 nm . First, we noticed that the absorption spectra show an absorption band proportional to the quantity of rhodium introduced in the melt and that the shape of the curve is similar to the one of $\text{BaTiO}_3\text{:Rh}$. Concerning the saturation intensity, the influence of rhodium is positive. Indeed, the saturation intensity decreases from 360 mW/cm^2 for the undoped crystal to 90 mW/cm^2 for the 2000 ppm rhodium doped sample. Nevertheless 90 mW/cm^2 is still a high value

compared to the saturation intensity obtained in $\text{BaTiO}_3\text{:Rh}$ meaning that BCT:Rh will require higher power sources to be operated. The results concerning the effective trap density are surprising. First, the effective trap density of the undoped crystal is very high. Second, the effective trap density is the same for all BCT crystals with rhodium doping from 200 to 2000 ppm, with iron doping and without doping. The first point might indicate that a photorefractive impurity is present in large amount in the undoped crystal and is responsible for the large effective trap density. The second point might indicate that the quantity of this impurity is so large that it offsets the influence of rhodium on the effective trap density. It was also noted that the effective trap density of the undoped crystal was very large at 515 nm [11]. Furthermore, we pointed out that the absorption spectrum of the undoped crystal was proportionnal to the one of the iron doped crystal, so that the impurity contained in the BCT crystals might be iron. Nevertheless, this point is not clear and to have a better understanding, we will now try to modelize the charge transport in BCT:Rh at 850 nm.

2. CHARGE TRANSPORT SITUATION

Modelization of the charge transport is of prime importance as it enables the tailoring of the material properties, and will allow the future growth of optimized samples that will meet the requirement of the different applications.

2.1. The two-level model with three-charge states of rhodium is not valid

Different characteristic features of $\text{BaTiO}_3\text{:Rh}$ can be found in BCT:Rh. For example, the absorption spectra for BCT:Rh and $\text{BaTiO}_3\text{:Rh}$ are similar, with a large absorption band around 600 nm, which is attributed to rhodium. In the case of rhodium doping in BaTiO_3 , it has been established that a two-level model with three charge states of rhodium applies [1,2]. Furthermore, the one-level model is dismissed by the dependence of the effective trap density on the intensity. It is thus natural to consider that BCT:Rh follow a two-center model with three charge states of rhodium, like $\text{BaTiO}_3\text{:Rh}$.

As BCT is close to BaTiO_3 , we use the results obtained on $\text{BaTiO}_3\text{:Rh}$ as starting point. A complete study of $\text{BaTiO}_3\text{:Rh}$ was performed by N.Huot [2] at 1.06 μm . He determined the parameters of the model for $\text{BaTiO}_3\text{:Rh}$ at this wavelength :

- the total amount of rhodium in a 1000 ppm rhodium doped sample and the different parameters related to Rh^{4+} and Rh^{5+} :
- the photoionisation cross sections.
- the recombination rates.
- the thermal emission rates.

Among these parameters, only the photoionisation cross sections depend on the wavelength. To determine their values at 850 nm, we performed measurements on $\text{BaTiO}_3\text{:Rh}$ at 850 nm using the same experimental procedure and the same samples as in [2]. The determined photoionisation cross section of Rh^{4+} and Rh^{5+} at 850 nm rhodium parameters for $\text{BaTiO}_3\text{:Rh}$ at 850 nm are given in table 3 (second column).

This set of parameters was a basis to begin our simulations on BCT:Rh. We simulated numerically the dependence of the gain on the grating spacing for several amounts of rhodium (200 ppm, 1000 ppm and 2000 ppm) according to a two-level model with three charge states of rhodium. To estimate the total quantity of rhodium in the crystals, we assumed that it was proportional to the amount of rhodium in the melt. Furthermore, we supposed that the quantity of rhodium that entered a crystal of BCT was the same as for a crystal of BaTiO₃. As the rhodium parameters for BCT:Rh might be slightly different from the ones of BaTiO₃:Rh, we then varied each parameter around its initial value. But, whatever the parameters we chose, simulations indicated always a variation of the effective trap density with rhodium doping (Fig. 8a) which is much larger than the one observed experimentally (Fig.8b). Indeed, comparing both figures (8a and 8b), we see that the variation of the effective trap density (which corresponds to the variation of the position of the maximum of the curves) induced by a variation of the concentration of rhodium is much lower experimentally [$13\text{-}17\mu\text{m}^{-1}$] than on the simulated curves [$7\text{-}15\mu\text{m}^{-1}$].

Therefore, it seems that a two-level model with three charge states of rhodium cannot explain the photorefractive effect in BCT:Rh. This point is confirmed by the fact that the undoped and iron doped samples are sensitive at 850 nm and they exhibit the same large effective trap density as the rhodium doped samples. Indeed, it should indicate that an impurity is present in these crystals in large amount and is responsible for the photorefractive effect at this wavelength. As the effective trap density in the rhodium doped samples is independent of the quantity of rhodium and is the same as in the undoped and iron doped crystals, it seems possible that the same impurity is responsible for the photorefractive effect in all crystals. Several points tend to indicate that this impurity might be iron. First, the absorption spectra of the undoped and iron doped samples are proportionnal as mentionned in 1.1. Then, EPR measurements made at the University of Osnabrück indicated the presence of a large amount of iron in the rhodium doped crystals [12]. Finally, it is well known that BaTiO₃:Rh contains iron as major native impurity. But, opposite to BaTiO₃:Rh, where the presence of iron was neglected, it seems here necessary to take it into account to modelize the photorefractive effect in BCT:Rh. Indeed, the amount of iron in the BCT crystals seems to be much larger than in the BaTiO₃ crystals. It might be due to the growth temperature, which is 200°C higher for BCT, that favors the incorporation of iron in the crystal. That is why we think that the photorefractive model that best explains the photorefractive properties of BCT:Rh is a three-level model with two-level of rhodium and one level of iron.

2.2. Charge transport model in BCT:Rh

Here we briefly present the three-level model and the simulations, which have given a good accordance between the model and the experimental results.

2.2.a. Description of the three-level model :

We consider a three level model with two levels of rhodium ($\text{Rh}^{3+/4+}$, $\text{Rh}^{4+/5+}$) and one level of iron ($\text{Fe}^{3+/4+}$). When the crystal is illuminated, we have the following charge transfers :

$$\begin{cases} \text{Rh}^{5+} \leftrightarrow \text{Rh}^{4+} + \text{hole} \\ \text{Rh}^{4+} \leftrightarrow \text{Rh}^{3+} + \text{hole} \\ \text{Fe}^{4+} \leftrightarrow \text{Fe}^{3+} + \text{hole} \end{cases} \quad (9)$$

We consider Rh^{4+} and Fe^{4+} as neutral states of charges and we note them respectively N and N_F . Then Rh^{3+} and Fe^{3+} can be considered as recombination centers and Rh^{5+} as a generation center for holes. Their densities are respectively N^- , N_F^- and N^+ . The electrical neutrality of the crystal is achieved by shallow donor and acceptor densities N_D and N_A . The compensation density is noted $N_{da} = N_D - N_A$. p is the density of free holes in the valence band.

A schematic diagram of the structure of the dopant levels is shown in Fig.9.

The photoionization cross-sections are noted S , β are the thermal emission rates and γ the rates of recombination.

If N_T is the total density of rhodium and N_{TF} the total density of iron, we have :

$$\begin{cases} N_T = N^- + N + N^+ \\ N_{TF} = N_F^- + N_F \end{cases} \quad (10)$$

The rate equations are :

$$\begin{cases} \frac{\partial N^-}{\partial t} = (S^- I + \beta^-) N - \gamma^- p N^- \\ \frac{\partial N^+}{\partial t} = -(S^+ I + \beta^+) N^+ + \gamma^+ p N \\ \frac{\partial N_F^-}{\partial t} = (S_F I + \beta_F) N_F - \gamma_F p N_F^- \end{cases} \quad (11)$$

The current density and the continuity equation for holes are :

$$\text{div } \vec{j} = e \frac{\partial (N_{DA} + p + N^+ - N^- - N_F^-)}{\partial t} \quad (12)$$

$$\vec{j} = e p \mu \vec{E} - \mu k_B T \nabla p \quad (13)$$

Poisson's equation relates the electric field \vec{E} to the charge densities :

$$\text{div } \vec{E} = \frac{e}{\epsilon} (N_{DA} + p + N^+ - N^- - N_F^-) \quad (14)$$

Then, we perform a classical resolution of these equations under an interference pattern illumination $I = I_0 + \frac{1}{2} (m \cdot I_0 e^{ikx} + \text{c.c.})$ (with I_0 the total incident illumination and m the modulation of the fringes).

We write each quantity as $Q = Q_0 + \frac{1}{2}(Q_1 e^{ikx} + \text{c.c.})$, and linearize the equations, to obtain two sets of equations. The "zeroth-order" set relates the space-invariant quantities Q_0 to I_0 . The "first-order" set defines the amplitude of the induced space-charge field as a function of the modulation of the fringes. The sets are then solved numerically and we calculate the photorefractive gain as a function of the grating spacing and the pump intensity to compare with the experimental results.

2.2.b. Determination of the initial parameters

Following parameters are necessary for the simulations :

- rhodium parameters : the photoionisation cross-sections S^+ and S^- , the thermal ionization rates β_+ and β_- , the recombination rates γ^+ and γ^- and the total density of rhodium N_T .
- iron parameters : the photoionisation cross-section S_F , the thermal ionization rate β_F , the recombination rate γ_F and the total density of iron N_{TF} .
- the compensation density N_{da} .

For all the following numerical simulations, we supposed that the Fermi-level did not change with rhodium doping. So N_{da} was considered as a constant in all the crystals.

For the rhodium parameters, we used as initial values the $\text{BaTiO}_3\text{:Rh}$ ones determined at 850 nm (Table 3, second column).

To determine the iron parameters, we used the experimental results on the undoped crystal. Indeed, we considered that the photorefractive effect in this crystal was only due to iron. We simulated a charge transport model with one center ($\text{Fe}^{3+/4+}$). The initial values of the parameters were chosen according to [13], which gives indications about parameters for the iron level in $\text{BaTiO}_3\text{:Rh}$: β_F might be of the order of $5 \cdot 10^{-4} \text{ s}^{-1}$, S_F might be small compared to S^- and γ_F is not given and will be taken equal to γ^+ in $\text{BaTiO}_3\text{:Rh}$ (Table 3, second column). By adjusting the simulated curves to the experimental results, we obtained several informations about the values of the parameters. First, the dependence of the gain with intensity allowed us to precise both parameters S_F and β_F . Second, the value of the effective trap density gave us a minimum value for the quantity of iron contained in the crystal. Indeed, in the case of a one-center model, we have :

$$N_{\text{eff}} = N_F^- - \frac{N_{da}^2}{N_{TF}} = N_{da} \frac{(N_{TF} - N_{da})}{N_{TF}} \quad (15)$$

which can be written :

$$N_{da}^2 - N_{TF} N_{da} + N_{\text{eff}} N_{TF} = 0 \quad (16)$$

To have a solution of this second order equation, we should have :

$$N_{TF} (N_{TF} - 4N_{\text{eff}}) \geq 0 \text{ which corresponds to } N_{TF} \geq 4N_{\text{eff}}$$

And, knowing the value of the effective trap density ($6 \cdot 10^{22} \text{ m}^{-3}$), we have $N_{TF} \geq 24 \cdot 10^{22} \text{ m}^{-3}$.

When the value of N_{TF} is fixed, we obtained two values for the compensation density N_{da} : one is of the same order of magnitude as N_{TF} and the other is small compared to N_{TF} . Simulations showed that it is better to choose the smallest one according to experimental results.

2.2.c. Numerical simulations

With this initial set of parameters, we performed numerical simulations simultaneously on the undoped and rhodium doped samples for several amount of rhodium (200, 370, 600, 1000 and 2000 ppm). Our aim is to find a set of parameters which enables to explain our experimental results. To do this, we vary the parameters around their initial values to improve the accordance between simulated curves and experimental ones. It was possible to find a convenient set of parameters (Table 3, first column) so that the simulated curves were in good accordance with the experimental results (Fig.10). Figure 10a represents the dependence of the gain on intensity. The experimental points and the simulated curves are represented. We consider that we obtained a good accordance because a slight variation of some parameters induces a large translation of the curves. In figure 10b, we have represented the dependence of the gain with the grating spacing. The lines corresponds to simulations and we see that they are in good accordance with the experimental points

2.3. Validity of the model

Considering the number of unknown parameters, this set is certainly not the only one existing but the founded values seem to make sense. First, the values obtained for the rhodium parameters in BCT are not far from the ones obtained for $BaTiO_3:Rh$ at the same wavelength. Second, the large amount of iron that is found in BCT crystals is in good accordance with results obtained at the University of Osnabrück [12]. Moreover, it enables to explain the independence of the effective trap density on rhodium and iron doping. Indeed, it seems that the quantity of dopant added (rhodium or iron) is negligible compared to the amount of iron already present in the crystals because it does not change the effective trap density. But this point has to be further studied.

3. COMPARISON OF BCT:Rh WITH $BaTiO_3:Rh$

BCT:Rh is a good photorefractive material. Most of its photorefractive properties are not far from the ones of $BaTiO_3:Rh$.

The electrooptic coefficient r_{42}^T is really smaller than in $BaTiO_3$ but it still has a value that should be sufficient for applications such as phase conjugation. The ordinary photorefractive gain is of the same order of magnitude in both crystal. Indeed, the gain measured in a 2000 ppm $BaTiO_3:Rh$ (labelled X15) at 850 nm in a symmetrical configuration, with ordinary polarizations and for a grating wave vector of $3.6 \mu m^{-1}$ is around $1.1 cm^{-1}$ whereas it is around $1.3 cm^{-1}$ in a 2000 ppm BCT:Rh (Fig.4b)

The smallest response time is measured for a 2000 ppm rhodium doped BCT with counterpropagating waves and was around 3s for an intensity of $1 W.cm^{-2}$. To compare with $BaTiO_3:Rh$, we measured 8 s for a 2000 ppm BCT:Rh with a grating wave vector of $4 \mu m^{-1}$ for $I = 1 W.cm^{-2}$. In a 2000 ppm $BaTiO_3:Rh$ (labelled Y32-B), we measured $\tau = 60 s$ for $I = 600 mW.cm^{-2}$ for the same grating spacing

and the same wavelength. As the response time in BaTiO₃:Rh is proportionnal to I^{-1} , we should have a response time of 36 s for $I = 1 \text{ W.cm}^{-2}$. Thus the response time is better in BCT:Rh than in BaTiO₃:Rh for the same amount of rhodium.

The saturation intensity is the unique parameter that differs a lot between both crystals. It is 100 to 1000 times higher in BCT:Rh than in BaTiO₃:Rh due to the influence of iron.

4. CONCLUSION

BCT:Rh crystals are sensitive in the infrared at 850 nm. Rhodium doping improves this sensitivity in term of saturation intensity but the obtained saturation intensities are still very high compared to the ones of BaTiO₃:Rh. Moreover, rhodium doping up to 2000 ppm does not improve the effective trap density. It might be due to the presence of a large amount of iron in the crystals. Indeed, to explain the experimental results, we need to take into account the presence of iron in our charge transport model. Iron seems to be responsible for the high saturation intensity, what can be a drawback for some applications. To improve this characteristics of the BCT:Rh crystals, the next step in the developpement will be to test crystals doped with a larger amount of rhodium.

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TABLE CAPTIONS

Table 1 : Values of the saturation intensity, the effective trap density and the effective electrooptic coefficient determined for all studied crystals.

Table 2 : Dependence of the effective trap density and of the saturation coefficient η on intensity for a 1000 ppm rhodium doped BCT.

Table 3 : Set of parameters found to fit the experimental results in the case of a three-level model. The parameters of BaTiO₃:Rh determined at 850 nm according to a two-level model with three-charge states of rhodium are presented for comparison.

FIGURE CAPTIONS

Figure 1 : Absorption as a function of wavelength for BCT:Rh crystals doped with different amounts of rhodium.

Figure 2 : (a) Absorption spectra of BCT:Rh from which we have subtracted the absorption of the undoped sample and that we have divided by the amount of rhodium in the melt. (b) Absorption spectra for BCT:Fe and undoped BCT. The absorption spectra of the undoped crystal multiplied by 1.65 is almost identical to the absorption spectra of BCT:Fe.

Figure 3 : Experimental set-up for two-beam coupling experiments with copropagating waves; M : mirror, BS : beam splitter 50/50, H : hole, P : polarizer, D : detector.

Figure 4 : (a) Normalized photorefractive gain as a function of intensity. (b) Photorefractive gain in ordinary polarization as a function of the grating wave vector. In both figures, the points correspond to experiments and the lines corresponds to the fitting curves. ☒ undoped BCT, ■ BCT:Rh 200 ppm, ✱ BCT:Rh 370 ppm, ▲ BCT:Rh 600 ppm, ● BCT:Rh 1000 ppm, + BCT:Rh 2000 ppm, ⚡ BCT:Fe 140 ppm.







Figure 5 : Experimental set-up for the measurement of the gain with counterpropagating waves.

Figure 6 : Dependence of the gain with the grating wave vector for several laser intensities for a 1000 ppm rhodium doped BCT crystal.

Figure 7 : Experimental photorefractive gain Γ as a function of the angle between the grating wave vector and the c-axis around ($\beta = \pi/2$).

Figure 8 : (a) Effect of the variation of the amount of rhodium on the effective trap density in BCT:Rh, calculated according to a two-level model with three-charge states of rhodium. (b) Experimental dependence of the effective trap density on the amount of rhodium in BCT:Rh. The experimental points are represented as well as the fitting curves (the fit procedure is explained in 1.4.b). In both figures, the vertical lines show the maximum of the curves from which we deduce the effective trap density.

Figure 9 : Schematic energy level diagram of BCT:Rh containing iron as an impurity.

Figure 10 : Comparison between experimental points and simulated curves with simulations according to a three-level model with two levels of rhodium and one level of iron. Concerning the simulated curves of the gain as a function of the grating wave vector, we use in the calculations the value of r_{22}^{eff} experimentally determined for each crystal (see table 1). Figure 10 a represents the dependance of the normalized gain with intensity and figure 10 b concerns the dependence of the gain with the grating spacing.  undoped BCT,  BCT:Rh 200 ppm,  BCT:Rh 370 ppm,  BCT:Rh 600 ppm,  BCT:Rh 1000 ppm,  BCT:Rh 2000 ppm.

Crystal	I_{sat} (mW/cm ²)	N_{eff} (m ⁻³)	r_{22}^{eff} (pm/V)
BCT:Fe 140 ppm	1300	$5.4 \cdot 10^{22}$	15
undoped BCT	360	$6.2 \cdot 10^{22}$	11.6
BCT:Rh 200 ppm	190	$4.9 \cdot 10^{22}$	12.8
BCT:Rh 370 ppm	170	$6.1 \cdot 10^{22}$	13
BCT:Rh 600 ppm	110	$7.1 \cdot 10^{22}$	12.8
BCT:Rh 1000 ppm	100	$5.3 \cdot 10^{22}$	13.6
BCT:Rh 2000 ppm	90	$7.2 \cdot 10^{22}$	15.3

Table 1

Intensity (W/cm ²)	Effective trap density (m ⁻³)	$\eta(I)$
2	$6.5 \cdot 10^{22}$	1
0.2	$5.9 \cdot 10^{22}$	0.75
0.05	$4.1 \cdot 10^{22}$	0.54

Table 2

Parameters	Adjusted value for BCT:Rh	Adjusted value for BaTiO ₃ :Rh
Photoionization cross section of Rh ⁵⁺ : S ⁺ (m ²)	9e-21	1,24e-21
Photoionization cross section of Rh ⁴⁺ : S ⁻ (m ²)	0.4e-21	0,56e-21
Photoionization cross section of Fe ⁴⁺ : S _F (m ²)	0.001e-21	-
Recombination rate of Rh ⁵⁺ (m ³ /s) : γ_+	2.92e-14	3,5e-14
Recombination rate of Rh ⁴⁺ (m ³ /s) : γ_-	15* γ_+	1.5* γ_+
Recombination rate of Fe ⁴⁺ (m ³ /s) : γ_F	γ_+	–
Thermal emission rate of Rh ⁵⁺ (s ⁻¹) : β^+	2.3	2.3
Thermal emission rate of Rh ⁴⁺ (s ⁻¹) : β^-	0.0001	0,01
Thermal emission rate of Fe ⁴⁺ (s ⁻¹) : β_F	0.000125	-
Total Iron concentration (m ⁻³) : N _F	64e23	-
Compensation density (m ⁻³) : N _{DA}	0.82e23	-
Total Concentration of rhodium for a 1000 ppm doped sample(m ⁻³) : N _T	3.3e23	3.3e23

Table 3

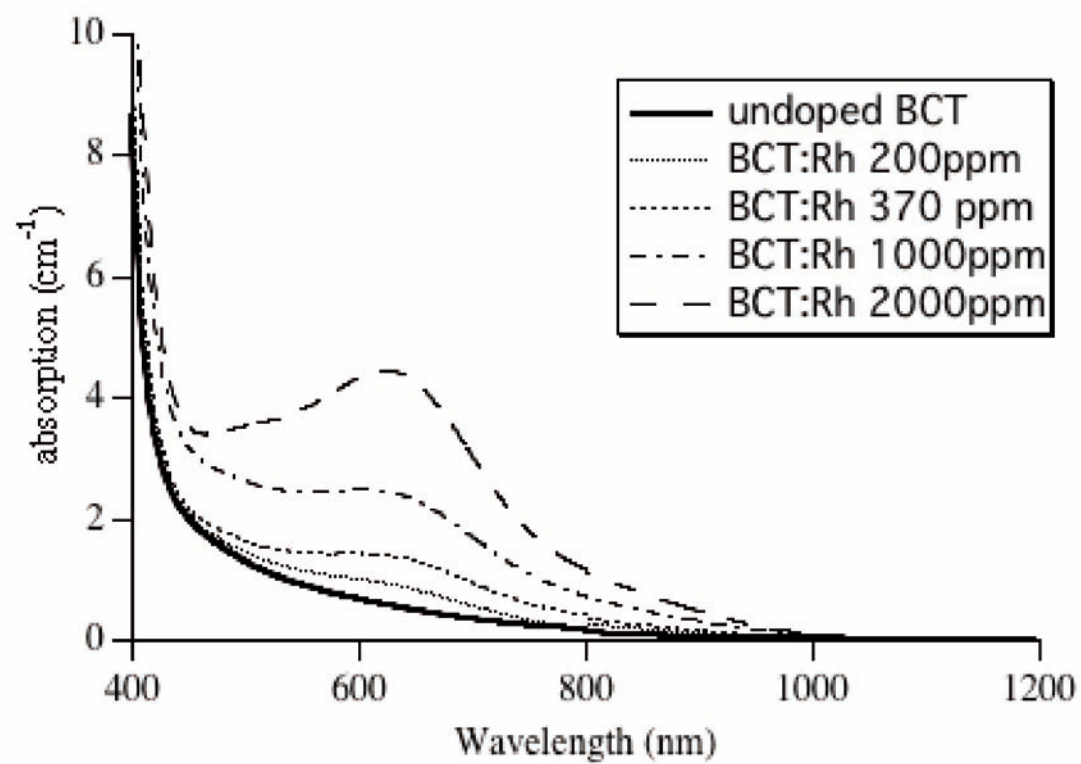


Figure 1

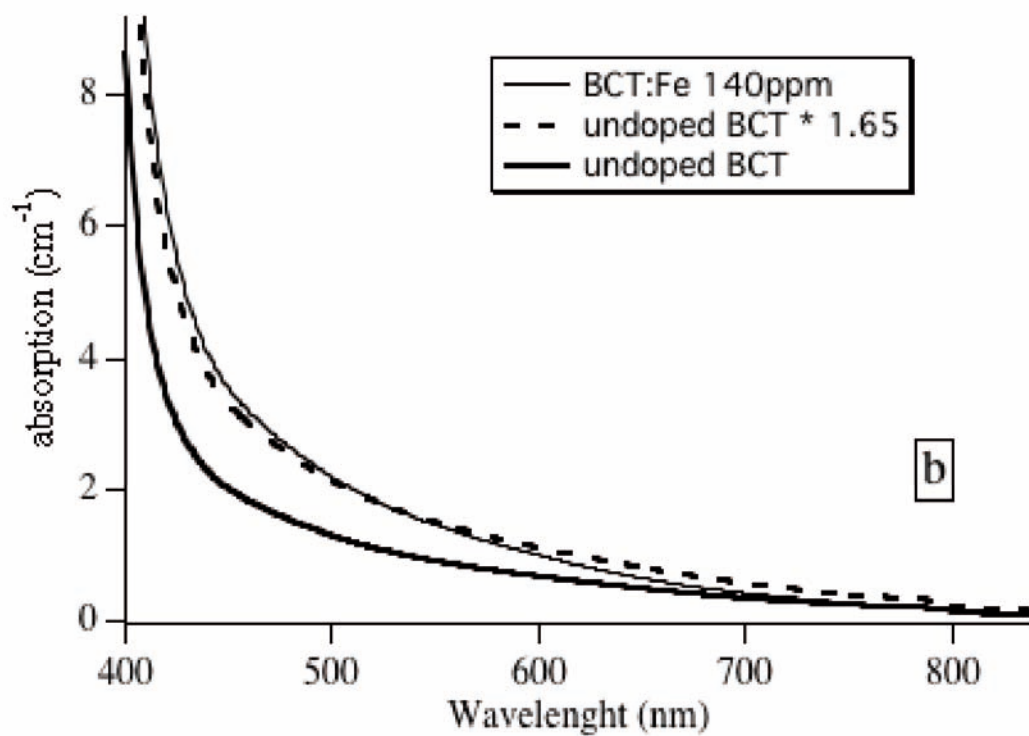
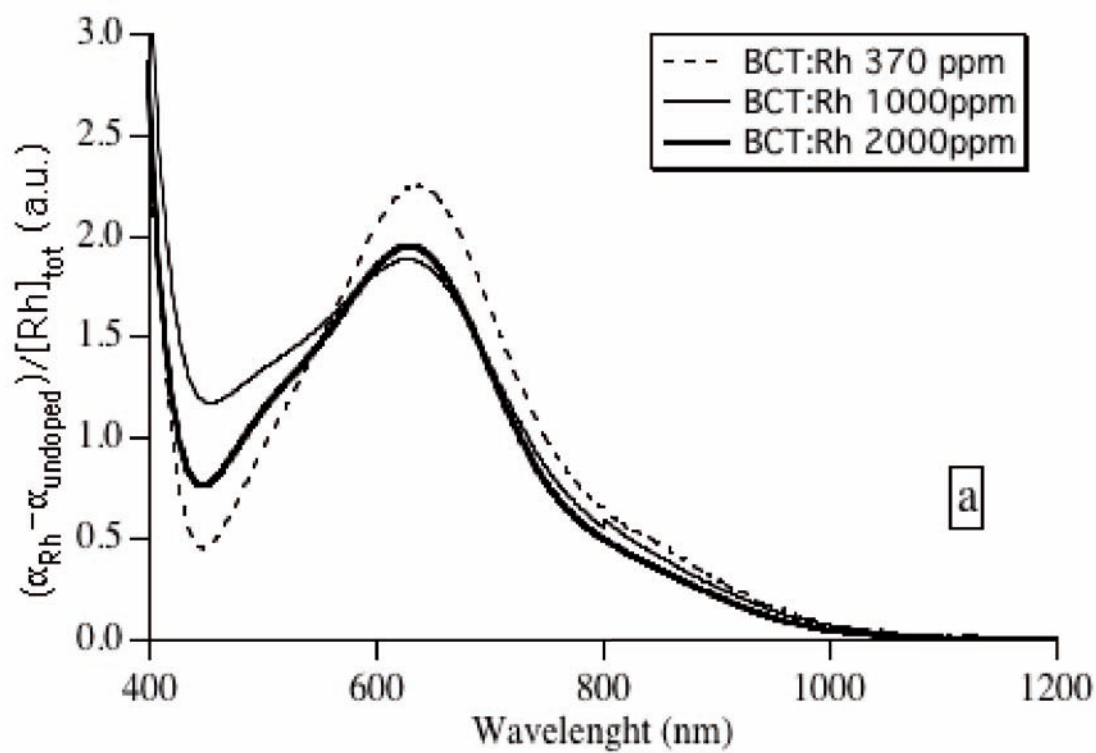


Figure 2

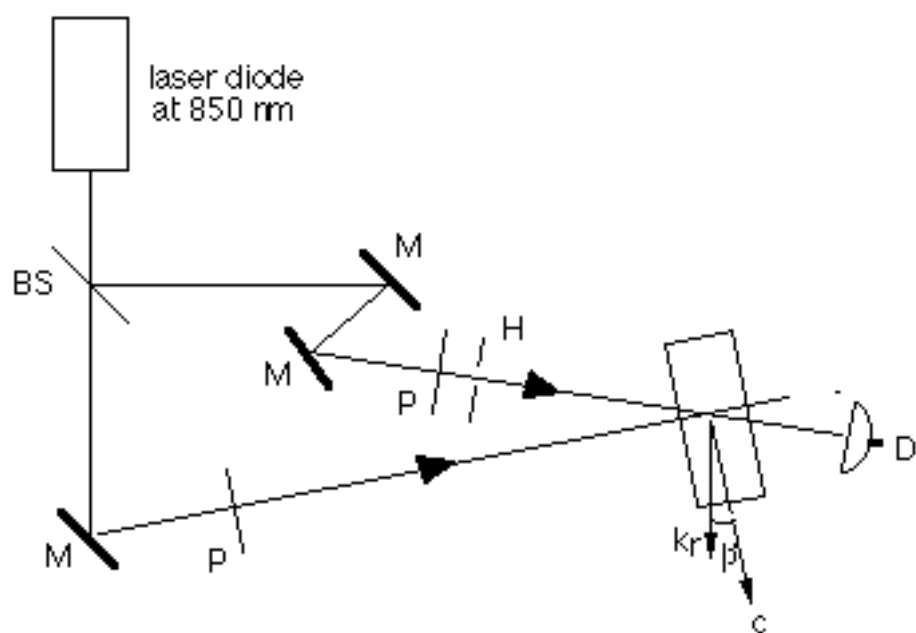


Figure 3

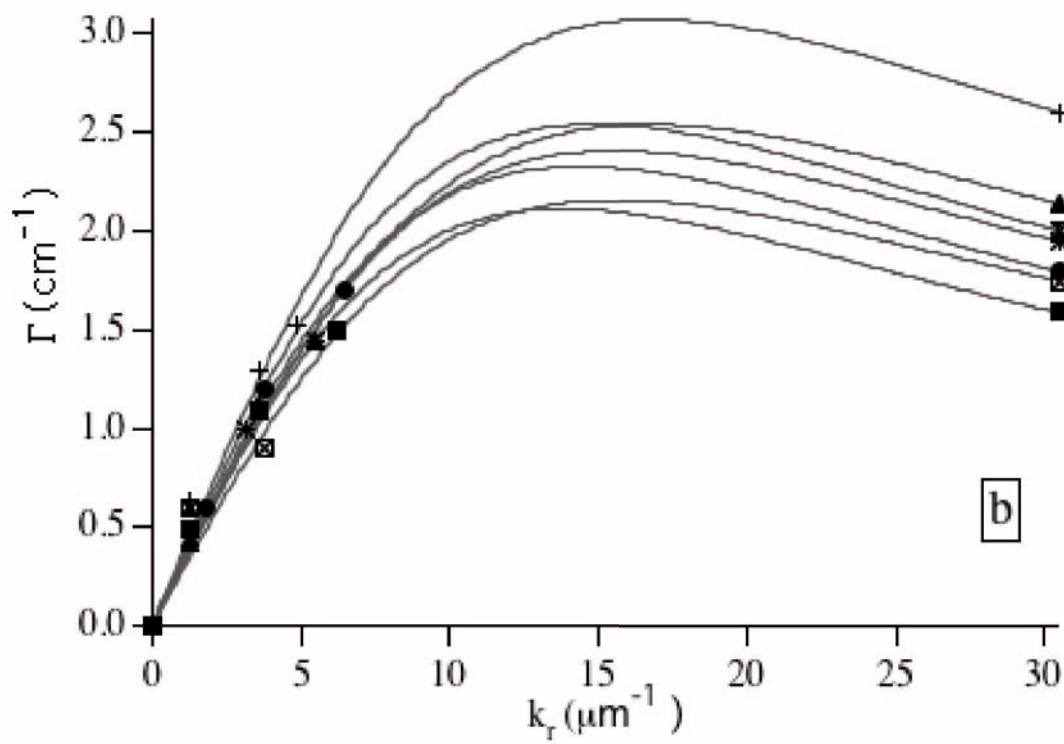
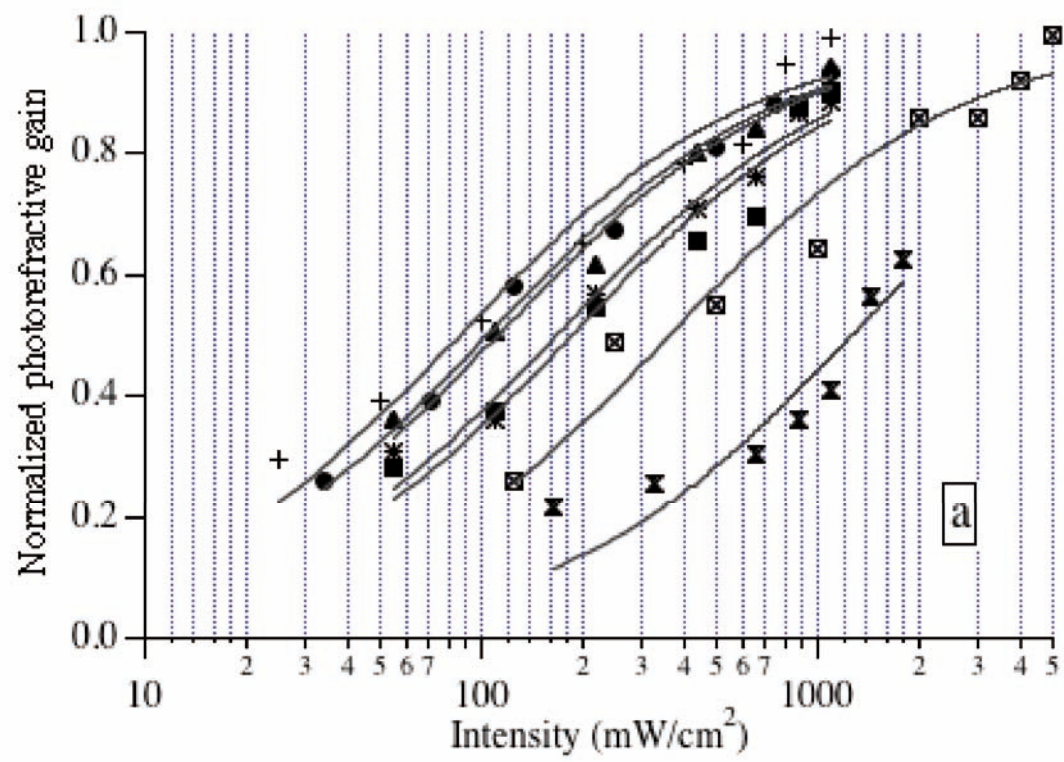


Figure 4

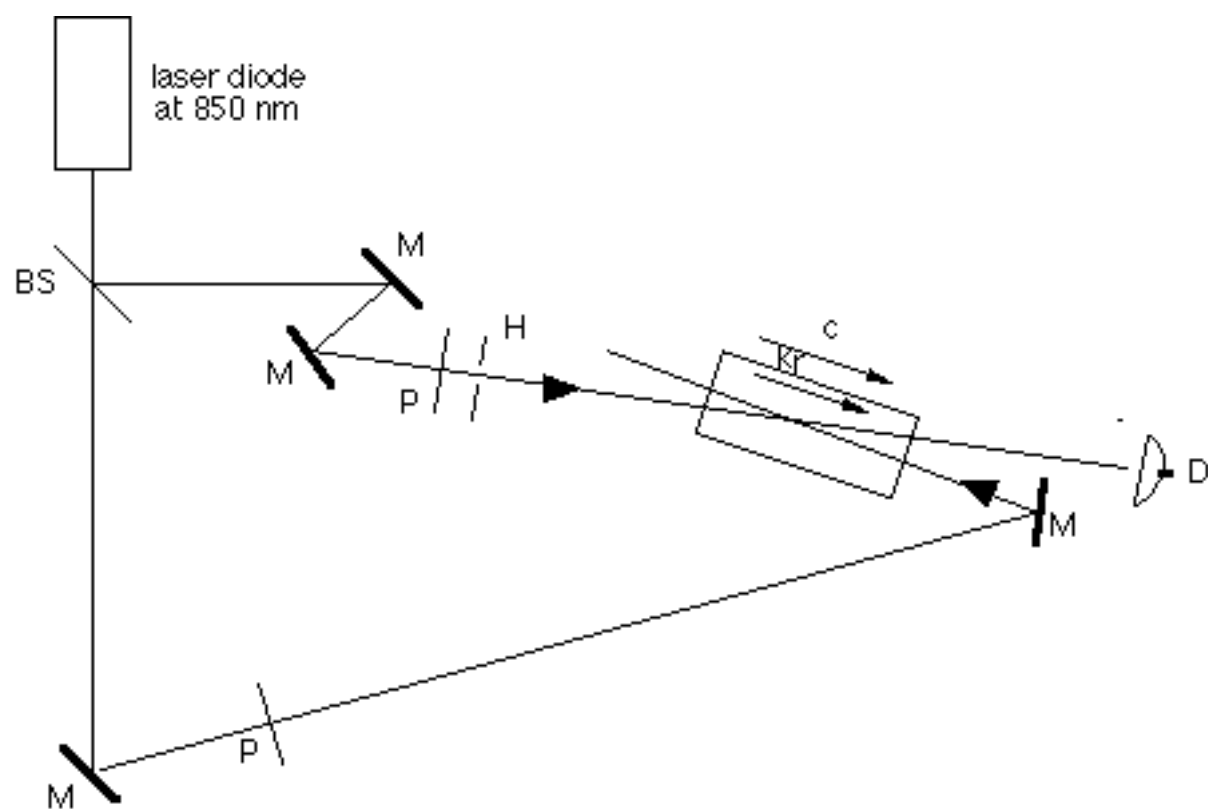


Figure 5

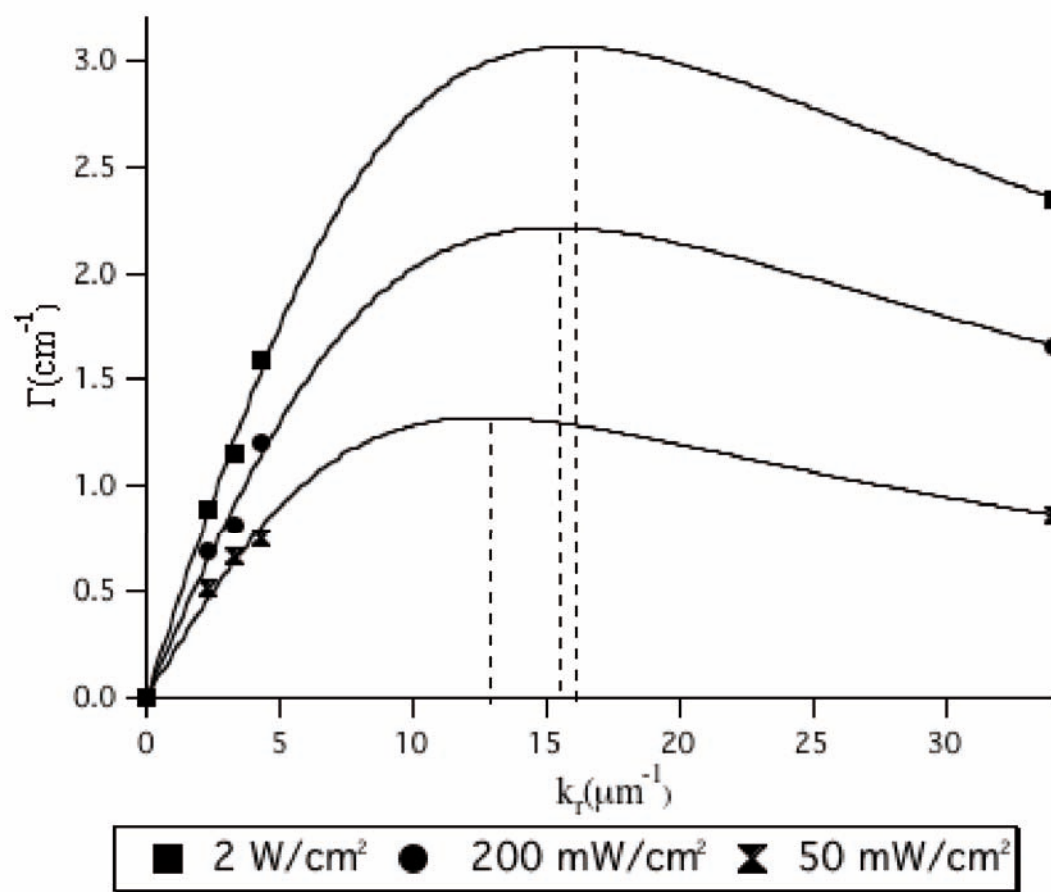


Figure 6

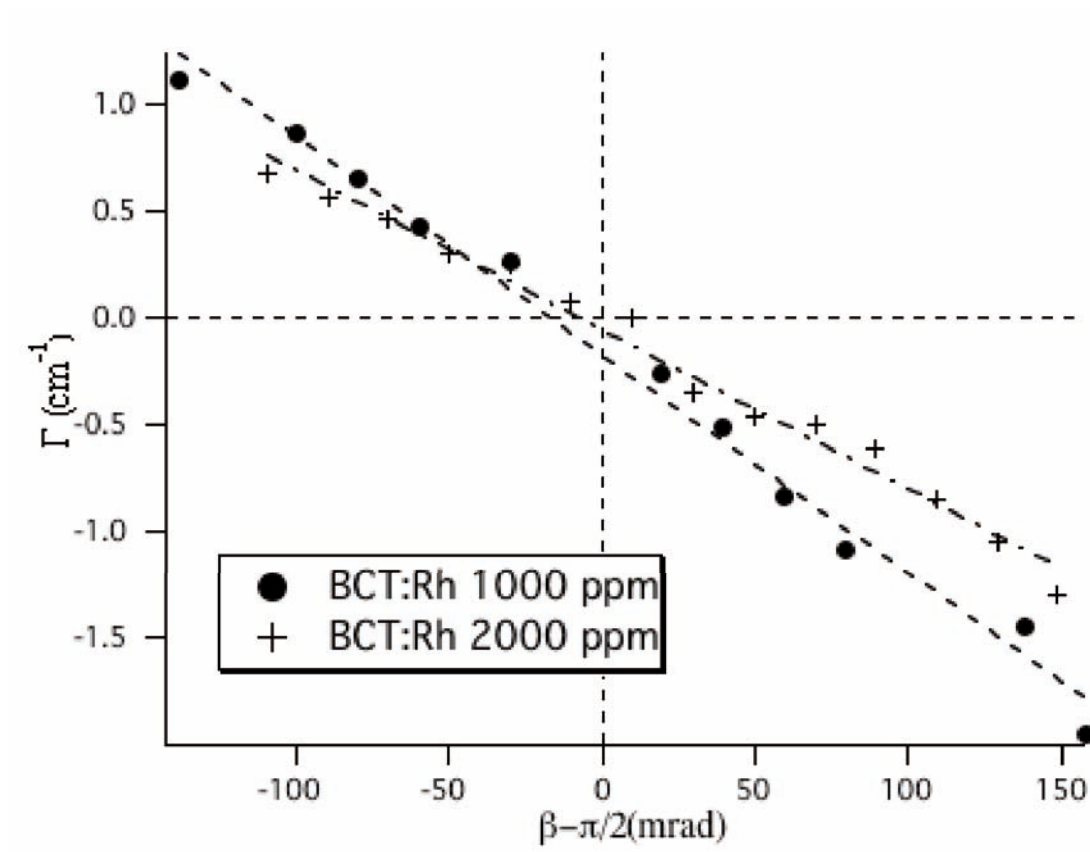


Figure 7

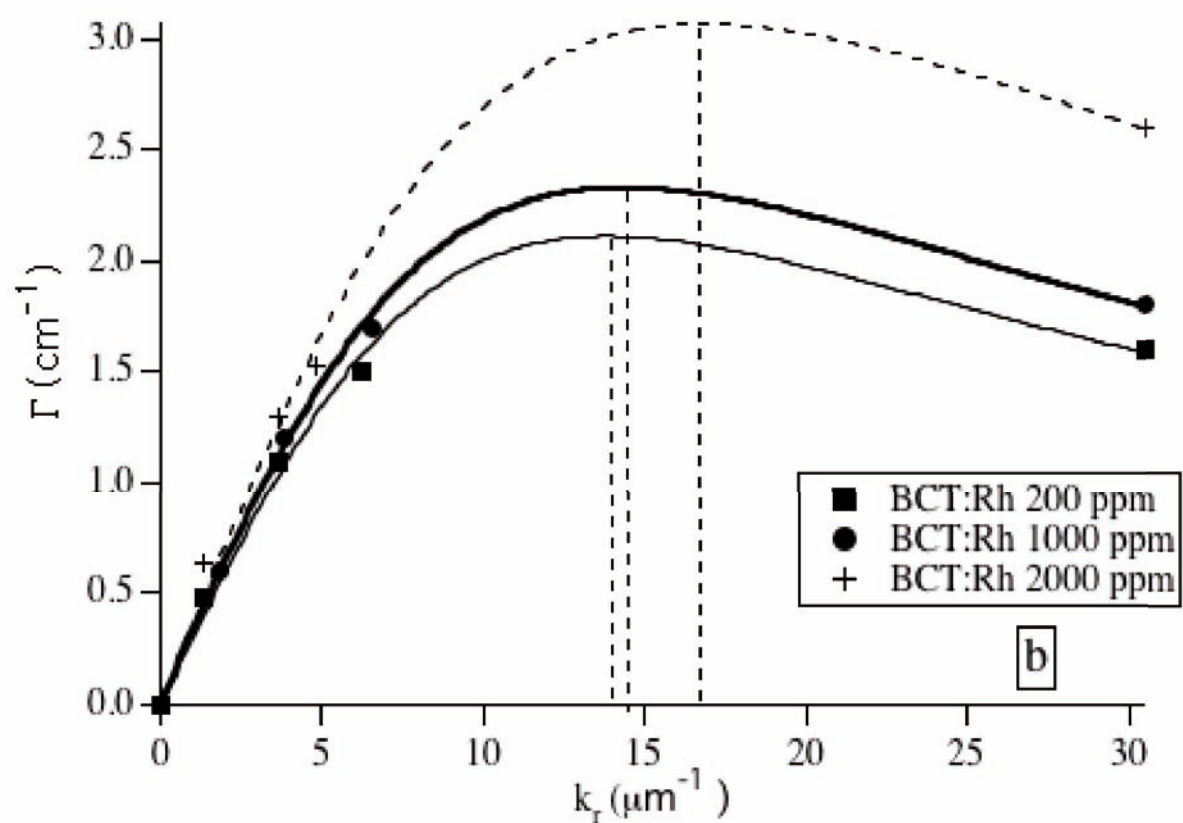
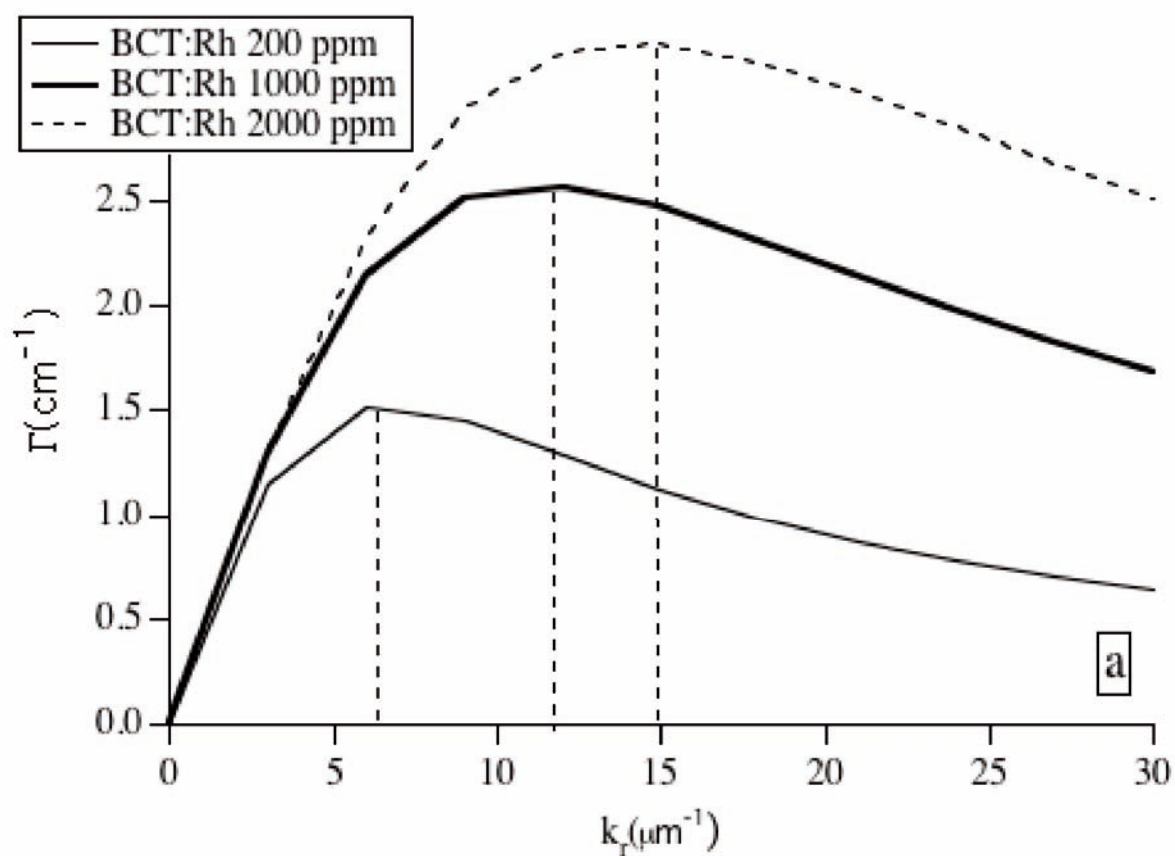


Figure 8

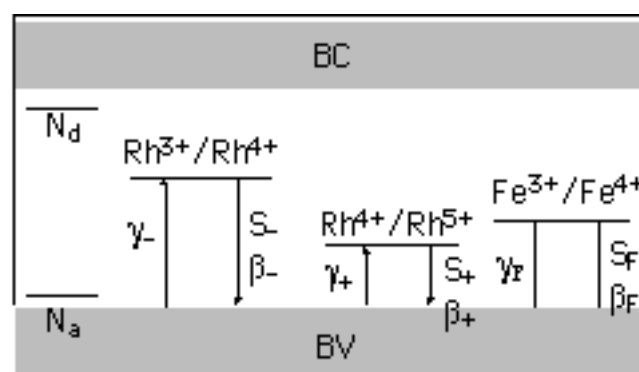


Figure 9

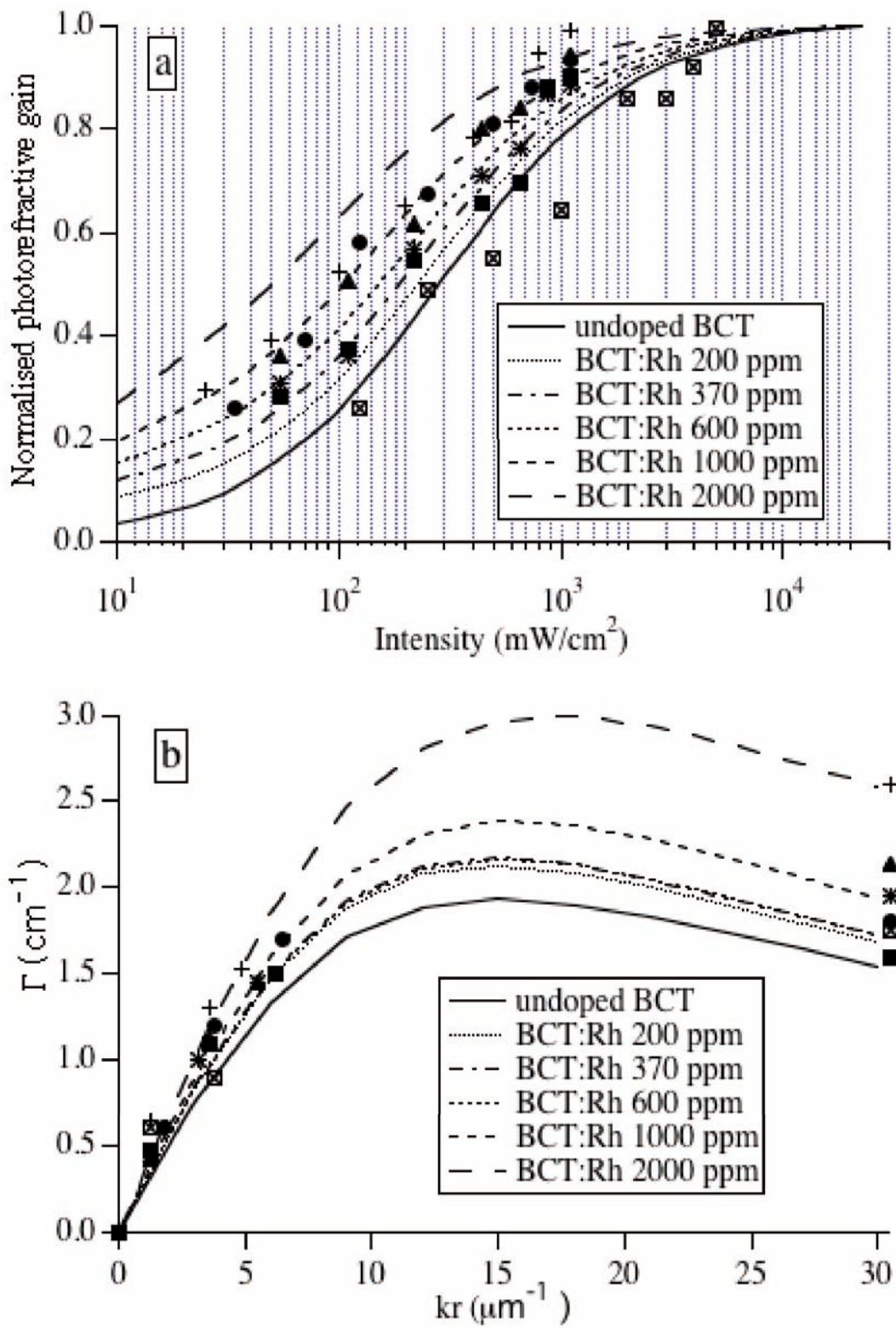


Figure 10